

CC 12, CHEMISTRY HONS.

1. Answer all the questions

[1×8=8]

- i. What is zero point energy.
- ii. Write the condition for normalization of wave function.
- iii. Write the wave function for antibonding orbital of H₂ molecule
- iv. What is electromagnetic radiation.
- v. Calculate degeneracy of the energy level having energy $14h^2/8ml^2$
- vi. What is the condition required for a molecule to be microwave active.
- vii. Define chemiluminescence.
- viii. Write the selection rule for vibrational spectra.

2. Answer any **Eight** the questions

[1.5×8=12]

- i. What are hot bands.
- ii. State Frank-Condon principle.
- iii. Calculate of degrees of freedom of CO₂.
- iv. What is stokes and anti-stokes line.
- v. Define P, Q, R branches.
- Vi. Give an example of photosensitized reaction.
- vii. Give an example of commutation rule.
- viii. Write the selection rule of mw spectra.
- ix. prblm force constant
- x. Write laws of photochemistry.

3. Answer any **Eight** the questions

[2×8=16]

- i. Define quenching and give an example.
- ii. Discuss fluorescence and phosphorescence process.
- iii. What is dissociation and predissociation process.
- iv. Define actinometry. Give an example of it.
- v. Write rule of mutual exclusion.
- vi. What is quantum yield. Give an example.
- vii. Discuss the reason behind low and high quantum yield.
- viii. Discuss Morse potential diagram regarding anharmonicity.

ix. prblm commutation

x. Bond length prblm

4. Answer any **Four** questions

[4×6=24]

i. Draw and discuss in brief regarding Jablonski diagram

ii. Discuss the methods for determination of quantum yield of a reaction.

iii. Discuss in brief regarding LCAO-MO treatment of H_2^+ .

iv. Discuss localized and non-localized molecular orbital treatment of H_2O .

V. Derive expression for calculation of energy a particle confined in an one dimensional box having edge length 'l'.

Vi. Derive expression for application of Schrodinger wave equation to a rigid rotor model of diatomic molecule.